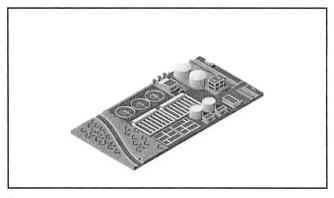


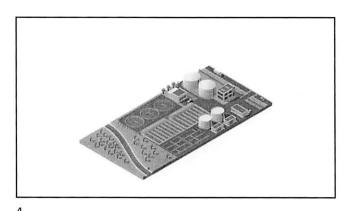
# Biological Metabolic Processes and Nutrient Removal

Presented by QUORYM
Innovator of Biological Treatment Systems

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# **Activated Sludge Process Control Parameters**

- DO
- PH
- Specific Oxygen Uptake Rate (SOUR)
- Mixed Liquor Suspended Solids
- Mixed Liquor Volatile Suspended Solids
- MLSS/MLVSS Ratio

Organic Compounds Removed

- Fat, Oil, and Grease (Fatty chain acids)
- BOD
- Nitrogen
- Phosphorus (Biological Uptake)

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# Organic Compounds Added

- · Sludge (Biomass)
- Oxygen
- · Media (Surface Area)

# Byproducts of Biological Removal

- N2 (Gas)
- CO2 (Gas)

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• H2S (In the absence of alternative e- acceptor)

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# Microbial Environment

- Protozoa
- Bacteria
  - Gram Positive and Gram Negative
- · Long Filamentous (Scaffolding)
- Enzymes (Biological byproduct)

#### Protozoa

- · Characterization evaluates microbial biodiversity
- Qualitative Analyses useful in determining control parameters
- · Not responsible for nutrient metabolism



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# Enzymes vs. Bacteria

- Enzymes
  - Act as a catalyst for redox reactions
     Not living, and are a biological byproduct
     Efficient at liquifying solids (ie. FOG)
     Similar to chemical treatment
- Will not metabolize pollutants
- Bacteria
   Uving organisms
   Produce enzymes to help break down solids and process carbon sources
   Bacteria will fully metabolize carbon sources

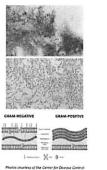
# **Bacteria Taxonomy**

- Gram Positive
  - Cell wall comprised of primarily Peptidoglycan
     No outer membrane

  - Stain purple in a Gram stain procedure

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- Gram Negative
   Cell wall comprised of outer membrane, periplasmic space, and cytoplasmic phospholipid membrane
   Stain red in Gram stain procedure



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# Types of Cellular Metabolism

#### Autotroph

- Does not require carbon source from environment
   Photoautotroph
- Requires Sunlight (anabolic pathway)
- Chemoautotroph
   Requires energy from inorganic oxidation (catabolic pathway)

- Requires Carbon source from environment
- Photoautotroph
   Requires Sunlight (anabolic pathway)
- · Chemoautotroph
  - Requires energy from inorganic oxidation (catabolic pathway)

### **Biochemical Properties**

#### Chemolithoheterotroph

- nittent DO levels (Aerobic and microaerobic concentrations, SBR process)

- Most efficient in intermittent DO Ievels (Aerobic and microaerobic concerts) of Grow rapidly in Aerobic environment
   Dentrification occurs in microaerobic and anaerobic conditions
   Capable of lipid metabolism
   i.e. nif-S-type betaproteobacteria, firmicutes, pseudomonas
- Chemolithoautotrophs

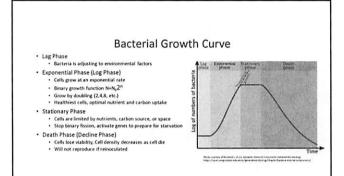
  - Typically obligate aerobes
    Convert ammonia to nitrate
    Not capable of fatty acid chain metabolism
    Only viable in liquid phase

  - Includes some Methogens, Halophiles, Sulfur Reducers, Nitrifiers, and Thermoacidophiles
     i.e. Nitrosomonas, Nitrobacter, etc.

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# **Biochemical Properties**

- · Anaerobic Chemoheterotrophs
  - Includes Methane producing bacteria
  - Includes Sulfide reducing bacteria
  - · Require carbon source
  - Obligate anaerobes (require <0.2 ppm DO)
- · Aerobic Chemoheterotrophs
  - · Highest rate of carbon source metabolism
  - Capable of lipid metabolism
  - · Abundant in aerobic treatment chambers



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#### Vegetative Cell Growth

- · Vegetative cells incapable of sporulation will follow cell growth
- · High uptake of nutrients
- · High uptake of carbon source
- Goal of continuous and fed-batch reactor growth is to maintain vegetative log state
- · Most gram negative cells or strictly vegetative
- Cannot be stored for long periods of time at room temperature (<2 weeks on Agar medium)

### Sporulation/Germination

- · Sporulation is a survival mechanism
- Ability to sporulate only expressed in Gram Negative
- · Induced by nutrient exhaustion

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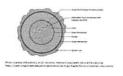
- · Produces highly resistant endospore · Sporulation induced at end of log phase
- After endospore is produced, nutrient uptake stops
- Cells capable of Sporulation ideal for dynamic systems
- with large nutrient range

 Upon environmental change suitable for vegetative state cells will germinate back to vegetative state

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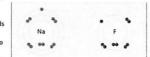
## EndoSpore

- · Highly resistant survival mechanism
- Resistant to a large range of:
  - Temperatures
  - Volume limitation
  - Nutrient depletion
- Capable of long storage time 1-3 months (@ room temp)
- Contains DNA to return back to vegetative state



What is Redox?

- Redox means "Reduction-Oxidation Potential"
- Two atoms (or molecules) that are unstable share an atom to increase the stability of the resulting molecule
- Unstable atom with an extra electron needs to give it away to be stable (Base)
- Unstable atom lacking an electron needs to take an electron to be stable (Acid)
- When they react the base oxidizes the acid, and the acid reduces the base



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#### What is Redox?

- . The "potential" for a redox reaction is dependent on the strength of the reactant
- Strong oxidizers (bases/caustics) like Sodium Hydroxide have a large potential
- . Strong acids (acids) like Sulfuric Acid have a large potential
- . The higher the potential of the reaction, the more heat will be produced
- Redox potential is similar to magnetism
  Low potential is comparable to a refrigerator magnet
  High potential is comparable to a rane lift magnet
  High potential will increase the "speed" that the reaction will take place

### Redox Biological Metabolic Processes

- Carbon Metabolism

   C,H<sub>11</sub>O<sub>4</sub> + 6O<sub>2</sub> to 6CO<sub>2</sub> + 6H<sub>2</sub>O + heat

   Nitrification

   2NH<sub>4</sub> \* + 3O<sub>2</sub> to 2NO<sub>2</sub> \* 4H\* + 2H<sub>2</sub>O

   2NO<sub>2</sub> \* + O<sub>2</sub> to 2NO<sub>3</sub>

   Denitrification

   NO<sub>1</sub> \* 2H\* + 2e\* to NO<sub>2</sub> \* + H<sub>2</sub>O

   NO<sub>2</sub> \* 2H\* 2e\* to NO + H<sub>2</sub>O

   NO<sub>2</sub> \* 2H\* 2e\* to NO + H<sub>2</sub>O

   NO<sub>2</sub> \* 2H\* 2e\* to NO + H<sub>2</sub>O

   NO<sub>2</sub> \* 2H\* 2e\* to N<sub>2</sub> + H<sub>2</sub>O

   Methanogenesis

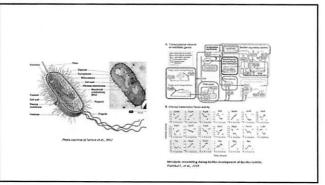
- Methanogenesis

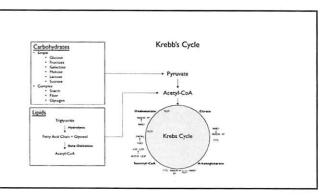
  CO<sub>2</sub> + 4H<sub>2</sub> to CH<sub>4</sub> + 2H<sub>2</sub>O (via Hydrogen gas partial pressure)

  CH<sub>2</sub>COOH to CH<sub>4</sub> + CO<sub>2</sub> (via acetate pathway)
- Sulfide Reduction
   SO<sub>4</sub><sup>2</sup> + 4H<sub>2</sub> to H<sub>2</sub>S + 2H<sub>2</sub>O + 2OH<sup>2</sup>

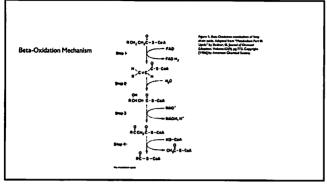
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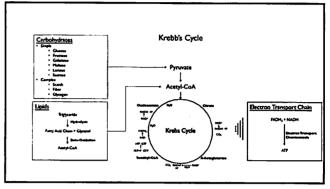
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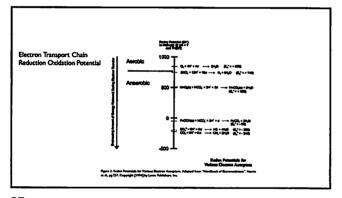
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Importance of Oxygen Transfer Oxygen availability is dependent only on the dissolved portion of oxygen Oxygen transfer is the amount of oxygen dissolved during an oxygen bubbles contact with water veters of oxygen transfer: Course air will have very low oxygen transfer Fine air will have very high oxygen transfer

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## **Benefits of Aerobic and Anaerobic Treatment Systems**

- robic (Activated Studge and Biofiltra Increased BOD rate of removal

- win runction better then anaerooic systems in low increased stability
   SSR systems have greater Nitrogen removal rates
   increased surface for blofilm production (blofiltratic

- nserobuc

   95% of Carbon is released as byproduct (CH4 and CO2) and only 0.05 contributes to bi

   Reduced studge waste

   Production of Methane for energy use

   Excellent pretreatment option for high poliutant wastewater concentrations

# **Solving Problems Using Cellular Metabolism**

- · Nitrogen (Ammonia, Nitrate)
- Phosphorus
- BOD
- FOG • H2S
- Settleability

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# Collection System Biological Activity

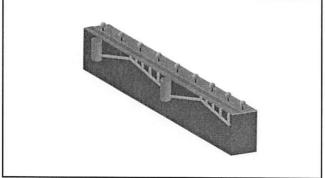
- Surface Area
- Dissolved Oxygen
- Nutrients (N & P)
- BOD
- FOG
- Detention Time
- Biofilm Biodiversity

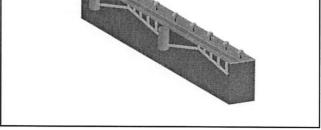
# Using Cellular Metabolism to Enhance Collection System Efficiency

- Biofilm Production
   Increase cell concentration for metabolic processes
- FOG (fatty acid chain) metabolism
   Reduces grease collection in sewer and lift stations
- BOD Metabolism
   Uptake of carbon sources for energy production
   Reduce Carbon source from potential Sulfide producing organisms
- Aerobic/Microaerophilic Conditions
   Introduce alternative electron acceptors for increased biodegradation and decrease Sulfide gas byproduct

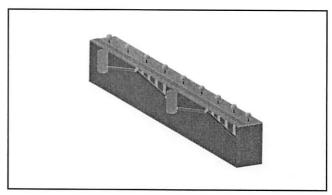
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# WWTP and Collection System Cell Troubleshooting

- Alternative electron acceptor (O2, N, Fe, MnO2)

  - Increase O2, N, Fe, MnO2
     Decrease Hydrogen Sulfide (H2S)

# WWTP and Collection System Cell Troubleshooting

- Cell Metabolism
  - · Increase Cell Concentration

  - Decrease N
     Decrease P (Bio P membrane uptake)

  - Decrease FOG

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# **WWTP and Collection System Cell Troubleshooting**

- Nitrifier
  - Increase Nitrifier
    - Increase Nitrate
    - Decrease Ammonia

# **WWTP and Collection System Cell Troubleshooting**

- Denitrifier
  - Increase Denitrifier
    - Increase N2 gas (Inert)
       Decrease Nitrate

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# **WWTP and Collection System Cell** Troubleshooting

- FOG
  - Decrease FOG
    - Increase Filamentous des
    - Increase Settleability

# **WWTP and Collection System Cell Troubleshooting**

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- Increase O2
  - Decrease denitrification
- Increase Metabolism · Increase Colony Growth
- Decrease O2
  - Increase denitrification
- Decrease Colony growth
   Remove Obligate Aerobes (<1ppm) Remove Obligate Anaerobes

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# **Biological Pretreatment Troubleshooting**

- · Carbohydrates/BOD
- Fatty Acid Chains/FOG
- Hydrogen Sulfide
- NBOD (Nitrogenous Biological Oxygen Demand)
- SSO (Sanitary sewer overflow)
- CSO (Combined sewer overflow)
- Nitrogen/Nitrate/Ammonia
- Corrosion

### References

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Questions?